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# Synthesis of Differently Protected 1-C-Methyl-Ribofuranoses Intermediates for the Preparation of Biologically Active 1'-C-Methyl-Ribonucleosides

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Starting from D-ribose, differently protected 1-C-methyl-D-ribofuranoses have been prepared as intermediates for the synthesis of variously modified 1'-C-methyl-ribonucleosides, a class of compounds potentially endowed with interesting biological activity.

Keywords 1-C-Methyl-D-ribofuranoses, 1-Deoxy-psicofuranoses, D-Ribonolactone, Tetrapropylammonium perruthenate

## **INTRODUCTION**

Synthetic nucleosides with modifications in the carbohydrate or base moiety are often characterized by anticancer<sup>[1]</sup> or antiviral<sup>[2–4]</sup> activity. In this respect, 40 years ago it was recognized that modifications at the 1'-position of adenosine, as in psicoadenosine (Fig. 1), "prevent deamination, while

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Figure 1: Structure of 1'-C-substituted nucleosides, 1-deoxypsicofuranose derivatives, and D-ribonolactone.

permitting good growth inhibitory activity."<sup>[5]</sup> Therefore, "modifications at this position, although chemically difficult, appear to be desirable."<sup>[5]</sup>

Due to this synthetic difficulty, only a few 1'-C-substituted nucleosides have been prepared so  $far^{[6-9]}$  and new promising biological activity for this class of compounds has been recently disclosed.[10] For our project related to the synthesis of analogues of 1'-C-methyl-adenosine,  $^{[11,12]}$  gram quantities of 1-deoxypsicofuranoses were required (Fig. 1).

## RESULTS AND DISCUSSION

## Synthesis of Differently Protected D-Ribonolactone Derivatives from D-Ribose

D-Ribonolactone would represent the ideal starting material for this synthesis, but its high price is not adequate to large-scale preparations. For this reason, D-ribose has been chosen as the alternative starting material as already proposed for the preparation of kilogram scale of a few D-ribonolactone derivatives.<sup>[13,14]</sup>

We have tried classical oxidations including the reactions with permanganate<sup>[13]</sup> or pyridinium chlorochromate<sup>[15]</sup> and the reagent of choice was tetrapropylammonium perrhuthenate (TPAP) under catalytic conditions with N-methylmorpholine N-oxide (NMMO) as co-oxidant.<sup>[16]</sup> This oxidizing system, previously applied to the synthesis of lactones from pyranoses and furanoses protected as benzyl and allyl ethers,  $[17]$  allowed a nearly quantitative transformation (95% of recovered purified product) of protected D-ribofuranoses 2, 3b, and 4 to the corresponding lactones 5, 6, and 7 (Sch. 1).

The preparation of lactone 6 started from the reaction of D-ribose (1a) with benzaldehyde in the presence of zinc chloride.<sup>[18]</sup> 2,3-Benzylidene derivative  $3a$ was obtained as an endo/exo mixture of the phenyl group of the benzylidene moiety, the  $\beta$ -anomer prevailing over the  $\alpha$ -one (8:1 ratio).<sup>[19]</sup> <sup>1</sup>H NMR



Scheme 1: Synthesis of protected D-ribonolactones 5, 6, and 7.

analysis of the two products purified by flash chromatography showed that the less polar product corresponds to the  $\beta$ -anomer, characterized by signals of H-C(1) (doublets at 5.53 and 5.56 ppm) coupled with H-C(2) ( $J < 1$  Hz). The more polar product corresponds to the  $\alpha$ -anomer, characterized by signals of H-C(1) (doublets at 5.96 and 5.98 ppm) coupled with H-C(2)  $(J = 3.7 \text{ Hz})$ . Compound **3a** was utilized as a crude  $\alpha/\beta$  mixture, silylated to ribofuranose **3b**, which was then oxidized with TPAP/NMMO to lactone 6 (3:1 exo/endo, overall yield from D-ribose 51%). The endo (or  $R$ ) configuration for the phenyl group of the major isomer of compound 6 was assigned by analysis of the NOE experiment. Saturation of the benzylidene proton resulted in NOEs of  $H-C(2)$  and  $H-C(3)$  signals, whereas  $H-C(4)$  was not affected. As a complementary result, in the minor isomer the irradiation of the benzylidene proton led to  $H-C(4)$  signal enhancement and the configuration *exo* (or S) was assigned.

We have also synthesized 2,3,5-tribenzyl-D-ribono-1,4-lactone 7, considering that the corresponding 1-C-methyl-ribofuranose 10 could be a useful intermediate for the synthesis of 1'-C-methyl-adenosine. 2,3,5-Tri-O-benzyl-D-ribofuranose (4) was prepared from 1-O-methyl-ribofuranose (1b) essentially as described in the literature<sup>[20]</sup> and quantitatively oxidized to lactone  $7 \text{ using }$ TPAP/NMMO (Sch. 1).

## Synthesis of 1-C-methyl-ribofuranoses

Reaction of protected D-ribonolactones with organometallic reagents is a well-known methodology for the synthesis of C-nucleosides.<sup>[21]</sup> It has been reported that ribonolactone 5 reacts with methyllithium at low temperature to

afford the corresponding 1-deoxy-psicofuranose  $8$ , as only one anomer.<sup>[6,7,11]</sup> Opening of the ribonolactone to the corresponding methyl ketone and related formation of a tertiary alcohol reported for other lactones<sup>[22]</sup> were not observed in the case of lactone  $5$ . [23] We have verified that 8 presents only the  $\alpha$ -OH and have not observed the formation of other competing side products. Under the same conditions, the methylation of 2,3-benzylidene lactone  $6^{[11]}$  afforded a mixture of unidentified products along with 1-deoxypsicofuranose 9 that could only be isolated in 38% yield. Compound 9 is an endo/exo mixture of the anomer that presents the methyl group in the  $\alpha$  position, as evidenced by NOE experiments in which irradiation of methyl group at C-1 led to H-C(5) enhancement. In the <sup>1</sup>H NMR spectrum of compound **9** we observed two signals corresponding to the 1-methyl at 1.62 ppm (exo) and 2.38 ppm (endo). The downshift of the signal at 2.38 ppm in the endo-9 can be explained by the influence of the vicinal phenyl group at the same side of the methyl group.

Finally, we examined the methylation of lactone 7 and obtained compound 10 in good yield (91% of purified product) as an anomeric mixture (1  $\alpha$ -CH<sub>3</sub>:3.5  $\beta$ -CH<sub>3</sub> ratio, as determined by NMR analysis) with no evidence for other side products (Sch. 2).

## Acetylation of 1-C-methyl-ribofuranoses

The synthesis of 1'-C-methyl-nucleosides required the acetylation of 1-deoxy-psicofuranoses before the coupling with the base.<sup>[6,7,9,11]</sup> The



Scheme 2: Methylation of lactones 5, 6, and 7.



Scheme 3: Acetylation of 1-C-methyl-ribofuranoses 8, 9, and 10.

acetylation of compound 8 has been studied in detail and it has been demonstrated that, starting from an anomerically pure 8, an  $\alpha/\beta$  mixture of acetate 11 is formed ( $\alpha$ -anomer 28%,  $\beta$ -anomer 58%).<sup>[11]</sup>

The acetylation of anomerically pure 9 yielded an equimolar mixture of four acetyl derivatives 12,  $\alpha$ -acetate (endo/exo), and  $\beta$ -acetate (endo/exo) with a small amount of acyclic compounds, and this discourages from the use of this intermediate for 1'-C-methyl-nucleosides synthesis. Under the same conditions of acetylation, 1-deoxy-psicofuranose 10 afforded only one product  $(60\% \text{ yield})$  that did not correspond to the expected cyclic acetate 14. <sup>1</sup>H NMR analysis of the isolated product showed resonances of H-C(5) at 5.32 ppm and of H-C(3) at 4.05 ppm, consistent with an acyclic structure. The singlet at 2.14 ppm in the  ${}^{1}H$  NMR spectrum and the signal at 208.19 in the 13C NMR spectrum suggested that a methyl ketone was present in the structure. Furthermore, the resonance at  $169.76$  ppm in the  $^{13}$ C NMR indicated the presence of an acetate and the structure of acyclic ketohexose was proposed for the product 13 (Sch. 3).

## **CONCLUSIONS**

We have investigated the preparation of a few 1-C-methyl-ribofuranoses starting from D-ribose (1a), which was transformed in the corresponding

derivatives 2, 3b, and 4. The oxidation of these compounds to the corresponding lactones 5, 6, and 7 has been quantitatively achieved with TPAP under catalytic conditions with NMMO as co-oxidant. The methylation of lactones 5 and 7 satisfactorily afforded 1-C-methyl-ribofuranoses 8 and 10, whereas the yield of 1-Cmethyl-ribofuranose 9 from lactone 6 was low. The acetates 11, 12, and 14 are key intermediates for the synthesis of 1'-C-methyl-nucleosides. Under standard acetylating conditions the 3,4-O-benzylidene derivative 9 gave the acetate mixture 12 in low yields and an open form was the only product obtained from the benzyl derivative 10. Only the acetate of 3,4-O-isopropylidene-1-C-methyl-ribofuranose 11 could be prepared in satisfactory yield.<sup>[12]</sup> In light of our results, starting from 1a, only the synthetic pathway leading to the acetates  $11$  is suitable for the synthesis of  $1'$ -C-methyl-nucleosides.

## EXPERIMENTAL

## General

Melting points were recorded on a Stuart Scientific SMP3 instrument and are uncorrected. Optical rotations were measured on a Perkin-Elmer 241 polarimeter using sodium lamp at 589 nm in CHCl<sub>3</sub>.<sup>1</sup>H NMR spectra were recorded at 303 K on Bruker AM-500 spectrometer operating at 500.13 and 125.76 MHz for <sup>1</sup>H and <sup>13</sup>C, respectively, equipped with an Aspect 3000 computer, a process control, and an array processor. The <sup>1</sup>H NMR chemical shifts are reported in parts per million, using as reference the signal for residual solvent protons  $(7.24$  for CDCl<sub>3</sub>), and coupling constants  $(J)$  are given in Hertz. In the  ${}^{13}$ C NMR spectra the residual solvent signal was used as an internal reference (CDCl<sub>3</sub>, triplet at  $\delta = 77.23$  ppm). All assignments were confirmed with the aid of two-dimensional  ${}^{1}H, {}^{1}H$  (COSYGPQF) or  ${}^{1}H, {}^{13}C$  (INV4GPQF) or  ${}^{1}H, {}^{13}C$  (INV4GPLRNDQF) experiments using standard Bruker pulse programs. Mass spectra were recorded on Finnigan LCQ-Deca (Termoquest) in ESI positive-ion mode, KV  $5.00$ ,  $225^{\circ}$ C,  $15$  V.

The progress of all reactions and column chromatography were monitored by TLC (Silica Gel 60  $F_{254}$  precoated plates with fluorescent indicator, Merck).

Purification of products was achieved by flash chromatography using silica gel 60 (230–400 mesh, Merck). 2,3,5-Tri-O-benzyl-D-ribofuranose  $(4)$ ,<sup>[20]</sup> 6-O-(tert-butyldimethylsilyl)-3,4-O-isopropylidene-1-deoxy-D-psicofuranose (8), and 2-O-acetyl-6-O-(tert-butyldimethylsilyl)-3,4-O-isopropylidene-1 deoxy-D-psicofuranose  $(11)^{[11]}$  were prepared according to literature and verified by NMR and ESI-MS.

#### 2,3-O-Benzylidene- $\alpha/\beta$ -D-ribofuranose (3a)

A mixture of D-ribose 1a  $(1.0 \text{ g}, 6.6 \text{ mmol})$ ,  $ZnCl<sub>2</sub> (0.83 \text{ g}, 6.1 \text{ mmol})$ , and benzaldehyde (4.5 mL, 45.0 mmol) was stirred for 18 h. Aqueous 40%

 $NaffSO<sub>3</sub>(75 mL)$  was added, the mixture stirred for 15 min, and, after evaporation at reduced pressure, the residue was treated with  $CH_2Cl_2$  (120 mL). The resulting slurry was stirred vigorously for 1 h and filtered and the filtrate evaporated. The residue was purified by flash chromatography (petroleum ether/ ethyl acetate, 6:4) and two products were separated corresponding to  $\alpha$ -anomer (Rf 0.125 in petroleum ether/ethyl acetate, 6:4) and  $\beta$ -anomer  $(Rf\ 0.375$  in petroleum ether/ethyl acetate, 6:4) (1:8) of the *endo/exo* mixture of title compound  $(64\% \text{ yield}).^{[19]}$ 

 $\beta$ -3a: 0.90 g; <sup>1</sup>H NMR endo  $\delta = 7.45 - 7.44$  (m, 5H, Ph), 5.80 (s, 1H, H-CPh), 5.56  $(d, J < 1, 1H, H-1), 4.90 (d, J = 6.2 Hz, 1H, H-2), 4.68 (d, J = 6.2 Hz, 1H, H-3),$ 4.58–4.57 (m, 1H, H-4), 3.82–3.75 (m, AB part of ABX system, 2H, H-5a,b) ppm; exo  $\delta = 7.39 - 7.36$  (m, 5H, Ph), 5.97 (s, 1H, H-CPh), 5.53 (d,  $J < 1$ , 1H, H-1), 4.98 (d,  $J = 5.6$  Hz, 1H, H-2), 4.69 (d,  $J = 5.6$  Hz, 1H, H-3), 4.53-4.50 (m, 1H, H-4), 3.82–3.75 (m, AB part of ABX system, 2H, H-5a,b) ppm.

 $\alpha$ -3a: 0.11 g; <sup>1</sup>H NMR endo  $\delta$  = 7.47–7.44 (m, 5H, Ph), 6.0 (s, 1H, H-CPh), 5.96  $(d, J = 3.7, 1H, H-1), 4.63$  (dd,  $J = 3.7$  Hz,  $J = 5.3$  Hz, 1H, H-2), 4.09 (dd,  $J = 5.3$  Hz,  $J = 8.4$  Hz, 1H, H-3), 3.97 (dd,  $J = 2.7$  Hz,  $J = 12.1$  Hz, 1H, H-5a), 3.91 (ddd,  $J = 2.7$  Hz,  $J = 2.7$  Hz,  $J = 8.4$  Hz, 1H, H-4), 3.73 (dd,  $J = 2.7$  Hz,  $J = 12.1$  Hz, 1H, H-5b) ppm; exo  $\delta = 7.41 - 7.38$  (m, 5H, Ph), 6.18 (s, 1H, H-CPh), 5.98 (d,  $J = 3.7$ , 1H, H-1), 4.71 (dd,  $J = 3.7$  Hz,  $J = 5.3$  Hz, 1H, H-2), 4.11 (dd,  $J = 5.3$  Hz,  $J = 8.4$  Hz, 1H, H-3), 3.97 (dd,  $J = 2.7$  Hz,  $J = 12.1$  Hz, 1H, H-5a), 3.91 (ddd,  $J = 2.7$  Hz,  $J = 2.7$  Hz,  $J = 8.4$  Hz, 1H, H-4), 3.73 (dd,  $J = 2.7$  Hz,  $J = 12.1$  Hz, 1H, H-5b) ppm.

#### 5-O-(tert-Butyldimethylsilyl)-2,3-O-benzylidene- $\alpha/\beta$ -D-ribofuranose (3b)

To a solution of the  $\alpha/\beta$  mixture of **3a** (1.0 g, 4.2 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub>  $(50 \text{ mL})$  under N<sub>2</sub>, triethylamine  $(0.64 \text{ mL}, 4.6 \text{ mmol})$ , DMAP  $(0.05 \text{ g},$ 0.42 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (5 mL) and tert-butyldimethylsilylchloride (0.69 g, 4.6 mmol) in dry  $CH_2Cl_2$  (30 mL) were sequentially added. The mixture was stirred at rt for 5 h and treated with 1M HCl solution to acidic pH, then with NaHCO<sub>3</sub> and H<sub>2</sub>O to pH 7.0. The solution was dried on  $Na<sub>2</sub>SO<sub>4</sub>$  and evaporated under reduced pressure to give crude  $\alpha/\beta$  mixture of 3b (0.90 g, 83% yield,  $endo$ /exo).<sup>[24]</sup>

 $\beta$ -3b: <sup>1</sup>H NMR endo  $\delta = 7.54 - 7.51$  (m, 2H, Ph), 7.43–7.38 (m, 3H, Ph), 5.79 (s, 1H, *H*-CPh), 5.47 (d,  $J < 1$ , 1H, H-1), 4.82 (d,  $J = 5.0$  Hz, 1H, H-2), 4.64 (d,  $J = 5.0$  Hz, 1H, H-3), 4.54 (dd,  $J = 2.4$  Hz,  $J = 2.4$  Hz, 1H, H-4), 3.88– 3.82 (m, AB part of ABX system, 2H, H-5a,b), 0.98 (s, 9H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.18 (s, 3H, SiCH<sub>3</sub>), 0.16 (s, 3H, SiCH<sub>3</sub>) ppm; exo  $\delta = 7.50 - 7.48$  (m, 2H, Ph), 7.54–7.51 (m, 2H, Ph), 6.00 (s, 1H, H-CPh), 5.44 (d,  $J < 1$ , 1H, H-1), 4.92  $(d, J = 5.0 \text{ Hz}, 1H, H-2), 4.66$   $(d, J = 5.0 \text{ Hz}, 1H, H-3), 4.49$   $(dd, J = 2.4 \text{ Hz},$   $J = 2.4$  Hz, 1H, H-4), 3.84–3.78 (m, AB part of ABX system, 2H, H-5a,b) 0.97  $(s, 9H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.19 (s, 3H, SiCH<sub>3</sub>), 0.18 (s, 3H, SiCH<sub>3</sub>) ppm.$ 

 $\alpha$ -3b: <sup>1</sup>H NMR endo  $\delta$  = 7.52–7.48 (m, 2H, Ph), 7.43–7.38 (m, 3H, Ph), 6.02 (s, 1H, H-CPh), 5.88 (d,  $J = 4.6$ , 1H, H-1), 4.50 (dd,  $J = 4.6$  Hz,  $J = 4.8$  Hz, 1H, H-2), 4.25 (dd,  $J = 4.8$  Hz,  $J = 5.3$  Hz, 1H, H-3), 3.95–3.90 (m, m 1H, H-4), 3.76–3.69 (m, AB part of ABX system, 2H, H-5a,b), 0.97 (s, 9H,  $\text{SiC}(\text{CH}_3)_3$ , 0.18 (s, 3H, SiCH<sub>3</sub>), 0.16 (s, 3H, SiCH<sub>3</sub>) ppm; exo  $\delta = 7.62-7.58$  $(m, 2H, Ph), 7.43-7.38$   $(m, 3H, Ph), 6.17$  (s, 1H, H-CPh), 5.91 (d,  $J = 4.6$ , 1H, H-1), 4.61 (dd,  $J = 4.6$  Hz,  $J = 4.8$  Hz, 1H, H-2), 4.23 (dd,  $J = 4.8$  Hz,  $J = 5.3$  Hz, 1H, H-3), 3.92–3.87 (m, 1H, H-4), 3.72–3.66 (m, AB part of ABX system, 2H, H-5a,b), 0.98 (s, 9H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.19 (s, 3H, SiCH<sub>3</sub>), 0.15 (s, 3H,  $SiCH<sub>3</sub>$ ) ppm.

## General Procedure for the Tetrapropylammonium Perruthenate (TPAP) Oxidation

A solution of ribofuranose derivative  $(5.0 \text{ mmol})$  in dry  $\text{CH}_2\text{Cl}_2$   $(40 \text{ mL})$ with NMMO  $(1.17 \text{ g}, 10.0 \text{ mmol})$  and molecular sieves  $(2.5 \text{ g})$  was stirred at rt and TPAP (0.175 g, 0.5 mmol) was added. After 30 min of stirring, the oxidation was complete and was filtered on a celite pad that was then washed with  $CH_2Cl_2$ . The filtrate was evaporated under reduced pressure to give the crude product.

#### 5-O-(tert-Butyldimethylsilyl)-2,3-O-isopropylidene-D-ribono-1,4-lactone (5)

Colorless crystals (1.44 g, 95%); mp 69–71°C [lit.<sup>[25]</sup> 69–70°C]; [ $\alpha$ ] $^{\rm 20}_{\rm D}$  –48.7 (c = 1.00, CHCl<sub>3</sub>) {<sup>[13]</sup>[ $\alpha$ ]<sub>D</sub><sup>20</sup> -46.6 (c = 0.80, CHCl<sub>3</sub>)}; <sup>1</sup>H NMR:  $\delta$  = 4.76-4.72 (m, 2H, H-2 and H-3), 4.61 (ddd,  $J = 1.0$  Hz,  $J = 1.9$  Hz,  $J = 5.6$  Hz, 1H, H-4), 3.91 (dd,  $J = 1.9$  Hz,  $J = 11.3$  Hz, 1H, H-5a), 3.82 (dd,  $J = 1.0$  Hz,  $J = 11.3$  Hz, 1H, H-5b), 1.49 (s, 3H, CH<sub>3</sub>), 1.40 (s, 3H, CH<sub>3</sub>), 0.89 (s, 3H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.09 (s, 3H, SiCH<sub>3</sub>), 0.07 (s, 3H, SiCH<sub>3</sub>) ppm. <sup>13</sup>C NMR  $(500 \text{ MHz}; \text{CDCl}_3, 25^{\circ}\text{C})$ :  $\delta = 174.17 \text{ (C=O)}$ ,  $112.99 \text{ (C<sub>out</sub>)}$ , 82.31 (C-4), 78.47  $(C-3)$ , 75.79  $(C-2)$ , 62.98  $(C-5)$ , 26.78  $(CCH_3)$ , 25.76  $(SiC(CH_3)_{3})$ , 25.58  $(CCH_3)$ , 18.19 (Si $C(CH_3)$ <sub>3</sub>),  $-5.62$  (SiCH<sub>3</sub>),  $-5.77$  (SiCH<sub>3</sub>) ppm.  $M/z$  320  $(M + NH_3)^+$ .

#### 5-O-(tert-Butyldimethylsilyl)-2,3-O-benzylidene-D-ribono-1,4-lactone (6)

Colorless syrup  $(1.66 \text{ g}, 95\%, endo/exo \text{ } 3:1)$ .

endo-6: <sup>1</sup>H NMR  $\delta$  = 7.49-7.47 (2H, m, Ph), 7.43-7.41 (m, 3H, Ph), 5.99 (s, 1H, PhCH),  $4.90-4.87$  (m, 2H, H-2 and H-3),  $4.77-4.76$  (ddd,  $J < 1.0$  Hz,  $J = 1.3$  Hz,  $J = 2.0$  Hz 1H, H-4), 3.96 (dd,  $J = 2.0$  Hz,  $J = 11.3$  Hz, 1H, H-5a), 3.87 (dd,  $J = 1.3$  Hz,  $J = 11.3$  Hz, 1H, H-5b), 0.93 (s, 9H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.12 (s, 3H, SiCH<sub>3</sub>), 0.11 (s, 3H, SiCH<sub>3</sub>) ppm. <sup>13</sup>C NMR:  $\delta = 172.95$  (C=O), 135.29

(PhC), 130.06 (PhCH), 128.52 (PhCH), 126.85 (PhCH), 106.57 (PhCH), 81.61  $(C-3)$ , 80.08  $(C-2)$ , 75.91  $(C-4)$ , 63,17  $(C-5)$ , 25.79  $(SiC(CH_3)_3)$ , 18.22  $(SiC(CH_3)_3)$ , –5.58  $(SiCH_3)$ , –5.72  $(SiCH_3)$  ppm.  $M/z$  373  $(M + Na)$ , 723  $(M + M + Na).$ 

exo-6: <sup>1</sup>H NMR  $\delta$  = 7.51–7.48 (2H, m, Ph), 7.43–7.42 (m, 3H, Ph), 5.98 (s, 1H, PhCH),  $4.99$  (d,  $J = 5.7$  Hz, 1H, H-2),  $4.82-4.80$  (m, 2H, H-3 and H-4),  $3.94$  (dd,  $J = 2.0$  Hz,  $J = 11.4$  Hz, 1H, H-5a), 3.87 (dd,  $J = 1.2$  Hz,  $J = 11.4$  Hz, 1H, H-5b), 0.91 (s, 9H,  $\text{SiC}(\text{CH}_3)_3$ ), 0.11 (s, 3H,  $\text{SiCH}_3$ ), 0.10 (s, 3H,  $\text{SiCH}_3$ ) ppm. <sup>13</sup>C NMR:  $\delta = 173.05$  (C=O), 135.70 (PhC), 130.05 (PhCH), 128.49 (PhCH), 126.47 (PhCH), 104.60 (PhCH), 84.13 (C-3), 77.86 (C-2), 76.42 (C-4), 63,18  $(C-5)$ , 25.77  $(SiC(CH_3)_3)$ , 18.21  $(SiC(CH_3)_3)$ ,  $-5.62$   $(SiCH_3)$ ,  $-5.72$   $(SiCH_3)$ ppm.  $M/z$  373 (M + Na), 723 (M + M + Na).

#### 2,3,5-O-Tribenzyl-D-ribono-1,4-lactone (7)

Colorless crystals (1.98 g, 95%); mp 53–54°C [lit.<sup>[26]</sup> 54–55°C]; [ $\alpha$ ] $_{{\rm D}}^{20}$  +73.6  $(c = 1.00, \text{ CHCl}_3)$  {lit.<sup>[27]</sup>  $[\alpha]_D^{20} + 74.1$   $(c = 2.00, \text{ CHCl}_3)$ }; <sup>1</sup>H NMR:  $\delta = 7.45-$ 7.30 (m, 13H, Ph), 7.24–7.17 (m, 2H, Ph), 4.98 (d,  $J = 11.9$  Hz, 1H, CHHPh), 4.78 (d,  $J = 11.9$  Hz, 1H, CHHPh), 4.73 (d,  $J = 11.9$  Hz, 1H, CHHPh), 4.60 (d,  $J = 11.9$  Hz, 1H, CHHPh), 4.57 (ddd,  $J = 1.6$  Hz,  $J = 2.4$  Hz,  $J = 2.7$  Hz, 1H, H-4), 4.52 (d,  $J = 11.0$  Hz, 1H, CHHPh), 4.45 (d,  $J = 5.9$  Hz, 1H, H-2), 4.44 (d,  $J = 11.0$  Hz, 1H, CHHPh), 4.14 (dd,  $J = 5.9$  Hz,  $J = 1.6$  Hz, 1H, H-3), 3.69 (dd,  $J = 2.7$  Hz,  $J = 11.0$  Hz, 1H, H-5a), 3.59 (dd,  $J = 2.4$  Hz,  $J = 11.0$  Hz, 1H, H-5b) ppm. <sup>13</sup>C NMR:  $\delta = 173.74$  (C=O), 137.30 (PhCH<sub>2</sub>), 137.19 (PhCH<sub>2</sub>), 173.01 (PhCH<sub>2</sub>), 128.54 (PhCH<sub>2</sub>), 128.28 (PhCH<sub>2</sub>), 128.14  $(PhCH<sub>2</sub>)$ , 128.08  $(PhCH<sub>2</sub>)$ , 128.02  $(PhCH<sub>2</sub>)$ , 127.63  $(PhCH<sub>2</sub>)$ , 81.82  $(C-4)$ , 75.43 (C-3), 73.78 (CH<sub>2</sub>), 73.68 (C-2), 72.76 (CH<sub>2</sub>), 72.42 (CH<sub>2</sub>), 68.80 (C-5) ppm.  $M/z$  441 (M + Na).

## General Procedure for the Methylation of D-Ribonolactone **Derivatives**

To a stirred solution of 1.6 M methyllithium in diethyl ether (2.13 mL) at  $-70^{\circ}$ C under argon atmosphere, a solution of D-ribonolactone derivative (2.0mmol) in anhydrous diethyl ether (20 mL) was added dropwise. After 1 h the mixture was warmed to 0°C, treated with 10% aqueous NH<sub>4</sub>Cl (20 mL) and extracted with diethyl ether  $(3 \times 20 \text{ mL})$ . The combined organic solutions were washed with ice-cold water  $(2 \times 15 \text{ mL})$  and dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ . Removal of the solvent under reduced pressure and purification of the residue by flash chromatography (petroleum ether/ethyl acetate, 9:1) afforded required 1-deoxy-psicofuranoside.

## 6-O-(tert-Butyldimethylsilyl)-3,4-O-benzylidene-1-deoxy-D-psicofuranose (9)

Colorless syrup  $(0.28 \text{ g}, 38\%, 3:1 \text{ endo}/\text{exo}).$ 

endo-9: <sup>1</sup>H NMR  $\delta$  = 7.53–7.50 (m, 2H, Ph), 7.43–7.40 (m, 3H, Ph), 5.94 (s, 1H, PhCH), 4.76 (d, 1H,  $J = 5.1$  Hz, H-3), 4.25 (dd,  $J = 5.1$  Hz,  $J = 7.1$  Hz, 1H, H-4), 3.85 (dd,  $J = 6.0$  Hz,  $J = 6.0$  Hz, 1H, H-6a), 3.79-3.75 (m, 2H, H-5 and H-6b), 2.38 (s, 3H, CH<sub>3</sub>), 0.93 (s, 9H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.11 (s, 3H, SiCH<sub>3</sub>), 0.10 (s, 3H,  $SiCH_3$ ) ppm.

exo-9: <sup>1</sup>H NMR  $\delta$  = 7.52–7.49 (m, 2H, Ph), 7.44–7.40 (m, 3H, Ph), 5.92 (s, 1H, PhCH), 5.05 (dd, 1H,  $J = 1.3$  Hz,  $J = 5.1$  Hz, H-4), 4.57 (d,  $J = 5.1$  Hz, 1H, H-3), 4.41 (ddd,  $J = 1.3$  Hz,  $J = 1.9$  Hz,  $J = 1.9$  Hz, 1H, H-5), 3.90–3.80 (m, 2H, H-6a and H-6b), 1.62 (s, 3H, CH<sub>3</sub>), 0.97 (s, 9H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.19 (s, 3H, SiCH3), 0.18 (s, 3H, SiCH3) ppm.

#### 3,4,6-Tribenzyl-1-deoxy-D-psicofuranose (10)

Colorless syrup  $(0.79 \text{ g}, 91\%$ , anomeric mixture 1: 3.5 ratio);<sup>[28]</sup> major isomer: <sup>1</sup>H NMR  $\delta = 7.43 - 7.35$  (m, 15H, Ph), 4.73 (d,  $J = 11.5$  Hz, 1H, CHHPh), 4.62 (d,  $J = 11.5$  Hz, 1H, CHHPh), 4.67–4.61 (AB system, 2H,  $CH_2Ph$ , 4.57 (d,  $J = 12.0$  Hz, 1H, CHHPh), 4.52 (d,  $J = 12.0$  Hz, 1H, CHHPh), 4.35 (ddd,  $J = 2.2$  Hz,  $J = 3.5$  Hz,  $J = 4.8$  Hz, 1H, H-5), 4.04 (dd,  $J = 2.2$  Hz,  $J = 5.1$  Hz, 1H, H-4), 3.76 (d,  $J = 5.1$  Hz, 1H, H-3), 3.54 (dd,  $J = 3.5$  Hz,  $J = 10.4$  Hz, 1H, H-6a), 3.48 (dd,  $J = 4.8$  Hz,  $J = 10.4$  Hz, 1H, H-6b), 1.51 (s, 1H, CH<sub>3</sub>) ppm. minor isomer: <sup>1</sup>H NMR  $\delta$  = 7.34–7.27 (m, 15H, Ph), 4.80 (d,  $J = 11.8$  Hz, 1H, CHHPh), 4.75 (d,  $J = 11.8$  Hz, 1H, CHHPh), 4.58 (d,  $J = 11.0$  Hz, 1H, CHHPh), 4.50 (d,  $J = 11.0$  Hz, 1H, CHHPh), 4.46– 4.40 (AB system, 2H, CH<sub>2</sub>Ph), 4.31–4.29 (m, 1H, H-5), 4.01 (dd,  $J = 2.4$  Hz,  $J = 4.4$  Hz, 1H, H-4), 3.84 (d,  $J = 4.4$  Hz, 1H, H-3), 3.68 (dd,  $J = 2.6$  Hz,  $J = 10.4$  Hz, 1H, H-6a), 3.65 (dd,  $J = 3.0$  Hz,  $J = 10.4$  Hz, 1H, H-6b), 1.58  $(s, 1H, CH<sub>3</sub>)$  ppm.

## General Procedure for the Acetylation of 1-Deoxy-Dpsicofuranoses

To a solution of 1-deoxy-psicofuranose (1.0 mmol) and DMAP (5 mg, 0.04 mmol) in pyridine  $(2 \text{ mL})$ , acetic anhydride  $(0.11 \text{ mL}, 1.0 \text{ mmol})$  was added dropwise and the reaction was stirred at rt for 5 h. Ice-cold water  $(5 \text{ mL})$  was then added and the aqueous phase was extracted with  $CHCl<sub>3</sub>$  $(3 \times 3$  mL). The combined organic layers were washed with a cold saturated NaHCO<sub>3</sub> solution ( $3 \times 3$  mL), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated under reduced pressure.

#### 5-O-Acetyl-3,4,6-tri-O-benzyl-1-deoxy-D-psicose (13)

From 3,4,6-tribenzyl-1-deoxy-D-psicofuranose (10), compound 13 was purified by flash chromatography (petroleum ether/ethyl acetate, 8:2);

colorless syrup (0.29 g, 60%);  $[\alpha]_D^{20} + 0.0$  ( $c = 1.00$ , CHCl<sub>3</sub>); <sup>1</sup>H NMR  $\delta = 7.36 -$ 7.24 (m, 15H, Ph), 5.32 (ddd,  $J = 4.0$  Hz,  $J = 4.0$  Hz,  $J = 6.8$  Hz, 1H, H-5), 4.65 (d,  $J = 11.9$  Hz, 1H, CHHPh), 4.62–4.59 (AB system, 2H, CH<sub>2</sub>Ph), 4.60 (d,  $J = 11.9$  Hz, 1H, CHHPh), 4.56 (d,  $J = 12.0$  Hz, 1H, CHHPh), 4.48  $(d, J = 12.0 \text{ Hz}, 1H, CHHPh), 4.21 (dd, J = 4.0 \text{ Hz}, J = 6.8 \text{ Hz}, 1H, H-4), 4.05$  $(d, J = 4.0 \text{ Hz}, 1H, H-3), 3.78-3.72 \text{ (m, 2H, H-6a,b), 2.14 (s, 3H, COCH<sub>3</sub>),$ 2.05 (s, 3H, OCOCH<sub>3</sub>) ppm. <sup>13</sup>C NMR:  $\delta = 208.19$  (C=O), 169.76 (C=O), 138.03 (PhC), 137.44 (PhC), 137.26 (PhC), 128.48 (PhCH), 128.42 (PhCH), 128.07 (PhCH), 127.99 (PhCH), 127.90 (PhCH), 127.74 (PhCH), 127.69  $(PhCH, 83.41(C-3), 78.43 (C-4), 73.25 (CH<sub>2</sub>), 73.17 (CH<sub>2</sub>), 72.92 (CH<sub>2</sub>), 71.35$  $(C-5)$ , 68.14  $(C-6)$ , 27.15  $(OCOCH_3)$ , 21.15  $(1-CH_3)$  ppm.  $M/z$  499  $(M + Na)$ .

Anal. Calcd for  $C_{29}H_{32}O_6$ : C, 73.09; H, 6.67. Found: C, 73.22; H, 6.74.

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#### **REFERENCES**

- [1] Bonate, P.L.; Arthaud, L.; Cantrell, W.R., Jr.; Stephenson, K.; Secrist III, J.A.; Weitman, S. Discovery and development of clofarabine: a nucleoside analogue for treating cancer. Nat. Rev. Dru. Discov. 2006, 5, 855–863.
- [2] De Clercq, E. Recent highlights in the development of new antiviral drugs. Curr. Opin. Microbiol. 2005, 8, 552–560.
- [3] Macchi, B.; Mastino, A. Pharmacological and biological aspects of basic research on nucleoside-based reverse transcriptase inhibitors. Pharmacol. Res. 2002, 46, 473–482.
- [4] Ren, J.; Stammers, D.K. HIV reverse transcriptase structures: designing new inhibitors and understanding mechanisms of drug resistance. Trend. Pharm. Sci.  $2005, 26, 4-7.$
- [5] Bloch, A.; Robins, M.J.; McCarthy, J.R., Jr. The role of the 5'-hydroxyl group of adenosine in determining substrate specificity for adenosine deaminase. J. Med. Chem. 1967, 10, 908–912.
- [6] Faivre-Buet, V.; Grouiller, A.; Descotes, G. Synthesis of thymine nucleosides derived from 1-deoxy-D-psicofuranose. Nucleos. Nucleot. Nucl. 1992, 11, 1651–1660.
- [7] Hayakawa, H.; Miyazawa, M.; Tanaka, H.; Miyasaka, T. A ribonolactone-based approach to the synthesis of 1'-carbon-substituted thymine ribonucleosides. Nucleos. Nucleot. Nucl. 1994, 13, 297–308.
- [8] Kodama, T.; Shuto, S.; Nomura, M.; Matsuda, A. An efficient method for the preparation of 1'alpha-branched-chain sugar pyrimidine ribonucleosides from uridine: the first conversion of a natural nucleoside into 1'-substituted ribonucleosides. Chem. Eur. J. 2001, 7, 2332–2340.
- [9] Cappellacci, L.; Barboni, G.; Palmieri, M.; Pasqualini, M.; Grifantini, M.; Costa, B.; Martini, C.; Franchetti, P. Ribose-modified nucleosides as ligands for adenosine receptors: synthesis, conformational analysis, and biological evaluation of 1'-Cmethyl adenosine analogues. J. Med. Chem. 2002, 45, 1196–1202.
- [10] Sommadossi, J.-P.; Gosselin, G.; Storer, R.; Egan, J. PCT Int. Appl. 2006, Application: WO 2005-US34786 20050926. Priority: US 2004–613085 20040924.
- [11] Ciuffreda, P.; Buzzi, B.; Alessandrini, L.; Santaniello, E. Activity of adenosine deaminase (ADA) and adenylate deaminase (AMPDA) towards 6-chloropurine nucleosides modified in the ribose moiety. Eur. J. Org. Chem. 2004, 4405–4409.
- [12] Ciuffreda, P.; Alessandrini, L.; Santaniello, E. 2', 3'-Isopropylidene group, a molecular scaffold to study the activity of adenosine and adenylate deaminase on adenosine analogues modified in the ribose moiety. Nucleos. Nucleot. Nucl. 2007, 26, 1311–1313.
- [13] Kaskar, B.; Heise, G.L.; Michalak, R.S.; Vishnuvajjala, B.R. A convenient large scale synthesis of protected D-ribonolactone from D-ribose. Synthesis 1990, 1031–1032.
- [14] Batra, H.; Moriarty, R.M.; Penmasta, R.; Sharma, V.; Stanciuc, G.; Staszewski, J.P.; Tuladhar, S.M.; Walsh, D.A. A concise, efficient and production-scale synthesis of a protected L-lyxonolactone derivative: an important aldonolactone core. Org. Process Res. Dev. 2006, 10, 484–486.
- [15] Liu, D.; Caperelli, C.A. A new synthesis of D-ribonolactone from D-ribose by pyridinium chlorochromate oxidation. Synthesis 1991, 933–934.
- [16] Ley, S.V.; Norman, J.; Griffith, W.P.; Marsden, S.P. Tetrapropylammonium per $r$ uthenate,  $Pr_4N^+RuO_4^-$ , TPAP: a catalytic oxidant for organic synthesis. Synthesis 1994, 639–666.
- [17] Benhaddou, R.; Czernecki, S.; Farid, W.; Ville, G.; Xie, J.; Zegar, A. Tetra-n-propylammonium tetra-oxoruthenate(VII): a reagent of choice for the oxidation of diversely protected glycopyranoses and glycofuranoses to lactones. Carbohydr. Res. 1994, 260, 243–250.
- [18] Wood, H.B.; Diehl, H.W.; Fletcher, H.G. Some products arising from the condensation of D-ribose with benzaldehyde. 2,3-O-benzylidene- $\beta$ -D-ribofuranose and di-(2,3-O-benzylidene-D-ribofuranose)-1,5':1',5-dianhydride. J. Am. Chem. Soc. 1956, 78, 4715–4717.
- [19] Grindley, T.B.; Szarek, W.A. Configurational and conformational studies on some benzylidene derivatives of D-ribose and di- $\beta$ -D-ribofuranose 1-5':1',5-dianhydride. Carbohydrate Res. 1972, 25, 187–195.
- [20] Barker, R.; Fletcher, H.G. 2,3,5-tri-O-benzyl-D-ribosyl and L-arabinosyl bromides. J. Org. Chem. 1961, 26, 4605–4609.
- [21] Wu, Q.; Simons, C. Synthetic methodologies for C-nucleosides. Synthesis 2004, 1533–1553.
- [22] Cavicchioli, S.; Savoia, D.; Trombini, C.; Umani-Ronchi, A. Syntheses of hydroxy ketones from lactones. J. Org. Chem. 1984, 49, 1246–1251.
- [23] Rodriguez, J.B. Chiral 1,4-dicarbonyl-2,3-O-isopropylidene derivatives. Rapid racemization on standing. Tetrahedron 1999, 55, 2157–2170.
- [24] Grochowski, E.; Stepowska, H. The synthesis of "C-O-N" analogues of nucleosides via the Mitsunobu reaction. Synthesis 1988, 795–797.
- [25] Cheng, J.C.-Y.; Hacksell, U.; Daves, G.D., Jr. Differentially Protected ribofuranoid glycals. J. Org. chem. 1985, 50, 2778–2780.

- [26] Timpe, W.; Dax, K.; Wolf, N.; Weidermann, H. 3-desoxyhex-2-enono-1,4-lactone aus D-hexofuran(osid)-urono-6,3-lactonen. Carbohydrate Res. 1975, 39, 53–60.
- [27] Jensen, H.S.; Limberg, G.; Pedersen, C. Benzylation of aldonolactones with benzyl trichloroacetimidate. Carbohydrate Res. 1997, 302, 109–112.
- [28] Sharma, G.V.M.; Chander, A.S.; Krishnudu, K.; Krishna, R.P. A simple and efficient PdCl<sub>2</sub> mediated conversion of  $\gamma$ ,  $\delta$ , -olefinic alcohols into C-glycosides. Tetrahedron Lett. 1997, 38, 9051–9054.